HEXAVALENT CHROMIUM (Cr VI) BY COLORIMETRIC METHOD SM 20 <sup>th</sup> Ed 3500-Cr B					
Facility Name:	VELAP ID				
Assessor Name:Analyst Name:		Iı	Inspection Date		
Relevant Aspect of Standards	Method Reference	Y	N	N/A	Comments
Records Examined: SOP Number/ Revision/ Date		Analyst:			
Sample ID: Date of Sample Preparation:		Date of Analysis:			
Hexavalent Chromium					
Were sample pH's adjusted to pH 9.3-9.7 using ammonium sulfate buffer solution, cooled to ≤ 6°C, and analyzed within 28 days? (The preservation temperature does not apply to samples analyzed within 15 minutes of collection.)	40 CFR 136.3 Table II Footnotes 18, 20				
Chromium					
If total chromium was determined, was the sample acidified with $HNO_3$ upon collection to a pH < 2?	3500-Cr A 3 Intro				
If dissolved chromium was determined, was the sample acidified to a pH < 2 with HNO $_3$ upon collection only after filtration through a 0.45 $\mu$ m filter?	3500-Cr A 3 Intro				
Were samples with turbidity NTU > 1 digested? (Applicable mainly to unfiltered samples.)	3030 E				
For sample digestion, was concentrated HNO <sub>3</sub> first added to a sample aliquot which was then brought to a slow boil?	3030 G 3				
For sample digestion, was concentrated H <sub>2</sub> SO <sub>4</sub> next added to the sample aliquot, and the sample aliquot heated until SO <sub>3</sub> fumes appeared?	3030 G 3				
Were sample aliquots next oxidized with potassium permanganate before reacting with diphenylcarbazide? (Necessary for samples with turbidity < 1 NTU as well.)	3500-Cr B 1				
Procedure					
Was spectrophotometer used at 540 nm, or did filter photometer have a maximum transmittance near 540 nm?	3500-Cr B 2 a				
Notes/Comments:					

## HEXAVALENT CHROMIUM (Cr VI) BY COLORIMETRIC METHOD SM 3500 B 20<sup>TH</sup> ED Page 2 of 3 Υ Ν N/A **Relevant Aspect of Standards** Method **Comments** Reference Was glassware acid washed, never exposed to 3500-Cr B 2 chromic acid, and free from scratches? Were standards carried through the same 3500-Cr B 4 preparations as samples? Were several drops of methyl orange indicator 3500-Cr B 4 followed by drops of ammonium hydroxide first added to samples until they turned yellow? Was H2SO4 added to samples next until they were 3500-Cr B 4 acidic? d 3500-Cr B 4 Were samples next heated to boiling? d Were drops of potassium permanganate added to 3500-Cr B 4 boiling samples until they turned red? d Were samples next boiled for 2 minutes longer 3500-Cr B 4 followed with addition of sodium azide solution until red color faded? Were samples finally boiled for 1 more minute? 3500-Cr B 4 After cooling, were portions of phosphoric acid 3500-Cr B 4 solution added to samples followed by adjustment of pH to $1.0 \pm 0.3$ with H2SO4? Was diphenylcarbazide solution added to samples 3500-Cr B 4 and the color allowed to develop? Were samples absorbances read, and their 3500-Cr B 4 absorbances corrected by the absorbance of a blank е carried through all preparations? Notes/Comments:

